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## **CLAIMS**

## What is claimed is:

- A composition comprising an aqueous dispersion of an electrically conductive organic polymer and a plurality of nanoparticles.
  - 2. A composition according to claim 1, wherein said electrically conductive organic polymer is selected from polyaniline with poly(2-acrylamido-2-methyl-1-propanesulfonic acid) as the counterion (PAni/PAAMPSA) and poly(ethylenedioxythiophene) with poly(styrenesulfonic acid) as the counter ion (PEDT/PSS).
  - 3. A composition according to claim 1, wherein said nanoparticles are inorganic nanoparticles.
  - 4. A composition according to claim 3, wherein said inorganic nanoparticles are selected from silica, alumina, and electrically conductive metal oxides.
  - 5. A composition according to claim 1, wherein said nanoparticles are organic nanoparticles.
  - 6. A composition according to claim 5, wherein said organic nanoparticles are selected from polyacrylates, carbon nanotubes, and perfluoroethylene sulfonates.
  - 7. A composition according to claim 1, wherein said nanoparticles have a particle size less than about 500 nm.
  - 8. A composition according to claim 1, wherein said nanoparticles have a particle size less than about 250 nm.
  - 9. A composition according to claim 1, wherein said nanoparticles have a particle size less than about 50 nm.
  - 10. A composition according to claim 4, wherein the weight ratio of silica:electrically conductive polymer is about 4:1.
- 11. A composition according to claim 4, wherein the weight ratio of electrically conductive oxides:electrically conductive polymer is about 1.5:1.
  - 12. A high resistance buffer layer comprising an electrically conductive polymer and a plurality of nanoparticles dispersed therein.
  - 13. A high resistance buffer layer according to claim 12, wherein said electrically conductive polymer is selected from PAni/PAAMPSA and PEDT/PSS.
    - 14. A high resistance buffer layer according to claim 12, wherein said nanoparticles are inorganic nanoparticles.

- 15. A high resistance buffer layer according to claim 12, wherein said inorganic nanoparticles are selected from silica, alumina, or electrically conductive metal oxides.
- 16. A high resistance buffer layer according to claim 12, wherein said nanoparticles are organic nanoparticles.

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- 17. A high resistance buffer layer according to claim 12, wherein said organic nanoparticles are selected from polyacrylates, carbon nanotubes, and perfluoroethylene sulfonates
- 18. A high resistance buffer layer according to claim 12, wherein said layer has a conductivity of less than about  $1 \times 10^{-3}$  S/cm.
- 19. A high resistance buffer layer according to claim 12, wherein said layer has a conductivity of less than about  $1 \times 10^{-5}$  S/cm.
- 20. An organic light emitting diode (OLED) comprising a high resistance buffer layer comprising an electrically conductive polymer and a plurality of nanoparticles dispersed therein.
- 21. An OLED according to claim 20, wherein said electrically conductive polymer is selected from PAni/PAAMPSA or PEDT/PSS.
- 22. An OLED according to claim 20, wherein said nanoparticles are inorganic nanoparticles.
- 23. An OLED according to claim 20, wherein said inorganic nanoparticles are selected from silica, alumina, or electrically conductive metal oxides.
  - 24. An OLED according to claim 20, wherein said nanoparticles are organic nanoparticles.
- 25. An OLED according to claim 20, wherein said organic nanoparticles are selected from polyacrylates, carbon nanotubes, and perfluoroethylene sulfonates.
  - 26. An OLED according to claim 20, wherein said buffer layer has a conductivity less than about  $1 \times 10^{-3}$  S/cm.
  - 27. A thin film field effect transistor electrode, comprising an electrically conductive polymer and a plurality of nanoparticles dispersed therein.
  - 28. A thin film field effect transistor electrode according to claim 36, wherein said electrically conductive polymer is selected from PAni/PAAMPSA or PEDT/PSS.
  - 29. A thin film field effect transistor electrode according to claim 27, wherein said nanoparticles are inorganic nanoparticles.

- 30. A thin film field effect transistor electrode according to claim 27, wherein said inorganic particles are metallic nanoparticles.
- 31. A thin film field effect transistor electrode according to claim 27, wherein said metallic nanoparticles are molybdenum nanoparticles.
- 32. A thin film field effect transistor electrode according to claim 27, wherein said nanoparticles are organic nanoparticles.
- 33. A thin film field effect transistor electrode according to claim 27, wherein said organic nanoparticles are carbon nanotubes.
- 34. A thin film field effect transistor comprising an electrode according to claim 27.

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- 35. A thin film field effect transistor according to claim 34, wherein said thin film field effect transistor has a conductivity greater than about 10 S/cm.
- 36. A method for reducing conductivity of an electrically conductive organic polymer film cast from aqueous dispersion onto a substrate to a value less than about 1 x 10<sup>-3</sup> S/cm, comprising adding a plurality of nanoparticles to said aqueous solution.
- 37. A method for producing buffer layers having increased thickness, comprising adding a plurality of nanoparticles to an aqueous dispersion of a conductive organic polymer, and casting a buffer layer from said aqueous dispersion onto a substrate.
- 38. A method for increasing conductivity of thin film field effect transistor electrodes cast from aqueous dispersion onto a substrate to a value greater than about 10<sup>-3</sup> S/cm, comprising adding a plurality of nanoparticles to said aqueous solution.
- 39. A method according to claim 38, wherein said nanoparticles are selected from metal nanoparticles and carbon nanotubes.